Age-Related Changes of Trace Element Contents in Intact Thyroid of Females Investigated by Neutron Activation Analysis

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Abstract

The prevalence of thyroid dysfunction is higher in the elderly as compared to the younger population. An excess or deficiency of trace element contents in thyroid plays an important role in goitrogenesis and carcinogenesis of gland. The variation with age of the mass fraction of silver (Ag), cobalt (Co), chromium (Cr), iron (Fe), mercury (Hg), rubidium (Rb), antimony (Sb), scandium (Sc), selenium (Se) and zinc (Zn) in intact (normal) thyroid of 33 females (mean age 54.5 years, range 3.5-87) was investigated by instrumental neutron activation analysis with high resolution spectrometry of long-lived radionuclides. Mean values ± standard error of mean for mass fractions (mg/kg, on dry-mass basis) of the trace elements studied were: Ag 0.0140±0.0020, Co 0.0505±0.0064, Cr 0.573±0.049, Fe 232±22, Hg 0.0329±0.0051, Rb 6.16±0.48, Sb 0.116±0.012, Sc 0.0042±0.0012, Se 2.22±0.23 and Zn 85.7±7.4. This work revealed that there is a significant increase in Co, Rb, Sb and Zn mass fraction in normal female thyroid during a lifespan. Therefore, a goitrogenic and carcinogenic effect of excessive Co, Fe, Rb, Sb and Zn level in the thyroid of old females should be studied.

Keywords: Age-related changes; Neutron activation analysis; Thyroid; Trace elements

Abbreviations

INAA-LLR: Instrumental Neutron Activation Analysis with high resolution spectrometry of Long-Lived Radionuclides
CRM/SRM: Certified/Standard Reference Materials
IAEA: International Atomic Energy Agency

Introduction

The endocrine organs, including the thyroid gland, undergo important functional changes during aging and a prevalence of thyroid dysfunction is higher in the elderly as compared to the younger population [1,2]. Advancing age is known to influence the formation of adenomatous goiter and thyroid cancer [3]. The prevalence of thyroid nodules is increased in the elderly, reaching a frequency of nearly 50% by the age of 65 [4]. Both prevalence and aggressiveness of thyroid cancer increase with age [2]. Women are affected by thyroid nodule and cancer two to five times more often than men [2-5].

Aging is a complex process involving biochemical and morphologic changes in single cells, in organs and in the whole organism. One of the most generally accepted explanations of how aging occurs at the molecular level is the oxidative stress hypothesis [6]. Reactive Oxygen Species (ROS) are widely considered to be a causal factor not only in aging but in a number of pathological conditions, including carcinogenesis. Aging, considered as an impairment of body functions over time, caused by the accumulation of molecular damage in DNA, proteins and lipids, is also characterized by an increase in intracellular oxidative stress due to the progressive decrease of the intracellular ROS scavenging [7]. Oxidative damage to cellular macromolecules which induce cancer can also arise through overproduction of ROS and faulty antioxidant and/or DNA repair mechanisms [8]. Overproduction of ROS is associated with inflammation, radiation and some other factors, including overload of some trace elements, in both blood and certain tissues, or deficiency of other trace elements with antioxidant properties [9-15]. Studies have shown that the imbalance in the composition of trace elements may cause different types of pathology. The importance of appropriate levels of many trace elements is indisputable, due to their beneficial roles when in specific concentrations, while on the other hand they can cause toxic effects with excessively high or low concentrations [12].

In our previous studies [16-24] the high mass fraction of Iodine (I) and some other trace element were observed in intact human thyroid gland when compared with their levels in non-thyroid soft tissues of the human body. However, some questions about the age-dependence of trace element mass fraction in thyroid of adult, and particularly, elderly females still remain unanswered. One valuable way to elucidate the situation is to compare the mass fractions of trace elements in young adult (the control group) with those in older adult and geriatric thyroid. The findings of the excess or deficiency of trace element contents in thyroid and the perturbations of their relative proportions in glands of adult and elderly females, may give an indication of their role in a higher prevalence of thyroid dysfunction in the elderly. The reliable data on trace element mass fractions in normal geriatric thyroid is apparently extremely limited. There are many studies regarding trace element content in human thyroid, using chemical techniques and instrumental methods [25-36]. However, the majority of these data are based on measurements of processed tissue and in many studies tissue samples are ashed (are burned in a muffle furnace) before analysis. In other cases, thyroid samples are treated with solvents (distilled water, ethanol etc.) and then are dried at a high temperature for many
hours. There is evidence that certain quantities of trace elements are lost as a result of such treatment [37-39]. Moreover, only a few of these studies employed quality control using Certified/Standard Reference Materials (CRM/SRM) for determination of the trace element mass fractions.

This work had three aims. The primary purpose of this study was to determine reliable values for the silver (Ag), cobalt (Co), chromium (Cr), iron (Fe), mercury (Hg), rubidium (Rb), antimony (Sb), scandium (Sc), selenium (Se) and zinc (Zn) mass fractions in the normal (intact) thyroid of subjects ranging from children to elderly females using non-destructive Instrumental Neutron Activation Analysis with high resolution spectrometry of Long-Lived Radionuclides (INAA-LLR). The second aim was to compare the Ag, Co, Cr, Fe, Hg, Rb, Sb, Sc, Se and Zn mass fractions in thyroid gland of age group 2 (adults and elderly persons aged 41 to 87 years), with those of group 1 (from 3.5 to 40 years) as well as to find the correlations between age and trace element contents and the final aim was to estimate the inter-correlations of trace elements in normal thyroid of females. All studies were approved by the Ethical Committee of the Medical Radiological Research Center, Obninsk, Russia (the reference no. 115050610007).

Materials and Methods

Samples of the human thyroid were obtained from randomly selected autopsy specimens of 33 females (European-Caucasian) aged 3.5 to 87 years within 48 hours after a sudden death. All the deceased were citizens of Obninsk and had undergone routine autopsy at the Forensic Medicine Department of City Hospital, Obninsk. Age ranges for subjects were divided into two age groups, with group 1, 3.5-40 years (30.9±3.1 years, M±SEM, n=11) and group 2, 41-87 years (66.3±2.7 years, M±SEM, n=22). These groups were selected to reflect the condition of thyroid tissue in the children, teenagers, young adults and first period of adult life (group 1) and in the second period of adult life as well as in old age (group 2). The available clinical data were reviewed for each subject. None of the subjects had a history of an intersex condition, endocrine disorder, or other chronic disease that could affect the normal development of the thyroid. None of the subjects were receiving medications or used any supplements known to affect thyroid trace element contents. The typical causes of sudden death of most of these subjects included trauma or suicide and also acute illness (cardiac insufficiency, stroke, embolism of pulmonary artery, alcohol poisoning). All right lobes of thyroid glands were divided into two portions using a titanium scalpel [40]. One tissue portion was reviewed by an anatomical pathologist while the other was used for the trace element content determination. A histological examination was used to control the age norm conformity as well as the unavailability of microadenomatosis and latent cancer.

After the samples intended for chemical element analysis were weighed, they were transferred to -20°C and stored until the day of transportation in the Medical Radiological Research Center, Obninsk, where all samples were freeze-dried and homogenized [41]. The pounded sample weighing about 50mg was used for trace element measurement by INAA-LLR. The samples for INAA-LLR were wrapped separately in a high-purity aluminum foil washed with rectified alcohol beforehand and placed in a nitric acid-washed quartz ampoule.

To determine contents of the elements by comparison with a known standard, Biological Synthetic Standards (BSS) prepared from phenol-formaldehyde resins were used [42]. In addition to BSS, aliquots of commercial, chemically pure compounds were also used as standards. Ten certified reference material IAEA H-4 (animal muscle) and IAEA HH-1 (human hair) sub-samples weighing about 50mg were treated and analyzed in the same conditions that thyroid samples to estimate the precision and accuracy of results. The vertical channel of nuclear reactor was applied to determine the content of Ag, Co, Cr, Fe, Hg, Rb, Sb, Sc, Se and Zn by INAA-LLR. The quartz ampoule with thyroid samples, standards and certified reference material was sealed, positioned in a transport aluminum container and exposed to a 24-hour neutron irradiation in a vertical channel with a neutron flux of 1.3 1013 ncm -2s -1. Ten days after irradiation samples were reweighed and repacked. The samples were measured for period from 10 to 30 days after irradiation. The duration of measurements was from 20 min to 10 hours subject to pulse counting rate. The gamma spectrometer included the 100 cm³ Ge(Li) detector and on-line computer-based MCA system. The spectrometer provided a resolution of 1.9 keV on the 60 Co 1332 keV line. Details of used nuclear reactions, radionuclides and gamma-energies were presented in our earlier publications concerning the INAA chemical element contents in human prostate and scalp hair [43,44].

A dedicated computer program for INAA mode optimization was used [45]. All thyroid samples were prepared in duplicate and mean values of trace element contents were used in final calculation. Using Microsoft Office Excel, a summary of the statistics, including, arithmetic mean, standard deviation, standard error of mean, minimum and maximum values, median, percentiles with 0.025 and 0.975 levels was calculated for trace element contents. The reliability of difference in the results between two age groups was evaluated by the parametric Student's t-test and non-parametric Wilcoxon-Mann-Whitney U-test. For the construction of "age - trace element mass fraction" diagrams and the estimation of the Pearson correlation coefficient between age and trace element mass fraction as well as between different trace elements the Microsoft Office Excel programs were also used.

Results

Table 1 depicts our data for Ag, Co, Cr, Fe, Hg, Rb, Sb, Sc, Se and Zn mass fractions in ten sub-samples of IAEA H-4 (animal muscle) and IAEA HH-1 (human hair) certified reference material and the certified values of this material. Table 2 represents certain statistical parameters (arithmetic mean, standard deviation, standard error of mean, minimal and maximal values, median, and percentiles with 0.025 and 0.975 levels) of the Ag, Co, Cr, Fe, Hg, Rb, Sc, Se and Zn mass fractions in intact (normal) thyroid of females. The comparison of our results with published data for the Ag, Co, Cr, Fe, Hg, Rb, Sb, Sc, Se and Zn contents in the human thyroid is shown in table 3. To estimate the effect of age on the trace element contents we examined two age groups, described above (Table 4). In addition, the Pearson correlation coefficient between age and trace element mass fraction was calculated (Table 5). Figure 1 shows the individual data sets for the Ag, Co, Cr, Fe, Hg, Rb, Sb, Sc, Se and Zn mass fraction in all samples of thyroid and also lines of trend with age. The data of inter-correlation calculations (values of r-coefficient of correlation) including all trace elements identified by us are presented in table 6.

Discussion

Good agreement of the Ag, Co, Cr, Fe, Hg, Rb, Sb, Sc, Se and Zn contents analyzed by INAA-LLR with the certified data of CRM IAEA H-4 and IAEA HH-1 (Table 1) indicates an acceptable accuracy of the results obtained in the study of trace elements of the thyroid presented in tables 2-5. The obtained values for Cr, Fe, Hg, Rb, Sc, Se and Zn...
contents, as shown in table 3, agree well with median of means cited by other researches for the human thyroid, including samples received from persons who died from different diseases. However, the means for Ag and Co are an order of magnitude lower than the median of previously reported data. A number of values for trace element mass fractions were not expressed on a dry mass basis by the authors of the
A significant age-related increase in Co, Fe, Rb, Sb and Zn mass fraction was observed in female thyroid (Table 4). In second group of females with mean age 66.3 years the mean mass fractions of these trace elements in thyroids were 1.42-1.96 times higher than in thyroids of the first age group (mean age 30.9 years). There were no statistically significant differences between the Ag, Cr, Hg, Sc and Se mass fractions within two different age-groups. Age-dependence of some trace element mass fractions found using the comparison between results for two age groups was confirmed for Co, Rb, Sb and Zn, while was not confirmed for Fe, when the Pearson correlation coefficient was calculated (Table 5). Moreover, the Pearson correlation coefficient showed a significant increase in Se mass fraction in female thyroid with age (Table 5).

The mass fractions of Co, Rb and Sb began to increase from the third decade and reached the highest values in the thyroid of elderly persons (Figure 1). The mass fraction of Fe increased in the third to sixth decades and reached a maximum at about the age of 50-60 years. After age 60 years, level of Fe began to decrease (Figure 1). This is the reason why the Pearson correlation did not show the age-dependence for Fe. The mass fraction of Zn increased in the third to seven decades and reached a maximum at about the age of 70 years. After age 70 years, content of Zn was maintained at more or less steady level (Figure 1). Age-dependence of Co, Fe and Zn mass fractions showed in present study did not agree with earlier findings. No other published data referring to age-related changes of Co, Fe, Rb, Sb and Zn mass fractions in human thyroid were found.

A significant direct correlation between the Co-Cr, Co-Rb, Co-Sc, Co-Se, Co-Zn, Fe-Hg, Rb-Sb, Sc-Se, Sc-Zn, Se-Zn mass fractions as well as an inverse correlation between Ag-Fe and Hg-Sb mass fractions was seen in female thyroid. No correlation was demonstrated between any other chemical elements (Table 6). If some correlations between the elements were predictable (e.g., Fe-Co), the interpretation of other observed relationships would require further study. No published data referring to inter-correlations Ag, Co, Cr, Fe, Hg, Rb, Sb, Sc, Se and Zn mass fractions in thyroid of females was found.

An age-related increase and excess in Co, Fe, Rb and Sb mass fractions in thyroid tissue may contribute to harmful effects on the thyroid.
There are good reasons for such speculations since many reviews and numerous papers raise the concern about toxicity and tumorigenesis of the metals [10,43,47-79]. Each of the metals is distinct in its primary mode of action. Moreover, there are several forms of synergistic action of the metals as a part of intracellular metabolism, during which several reactive intermediates and byproducts are created [47,48,53]. These reactive species are capable of potent and surprisingly selective activation of stress-signaling pathways, inhibition of DNA metabolism, repair and formation of DNA cross links, which are known to contribute to the development of human cancers [48,80,81]. In addition to genetic damage via both oxidative and non oxidative (DNA adducts) mechanisms, metals can also cause significant changes in DNA methylation and histone modifications, leading to alterations in gene expression [49,51,80]. *In vitro* and animal tumorigenic studies provided strong support for the idea that metals can also act as co-carcinogens in combination with nonmetal carcinogens [80]. All the deceased were citizens of Obninsk. Obninsk is the small nonindustrial city not far from Moscow in unpolluted area. None of those who died a sudden death had suffered from any systematic or chronic disorders before. The normal state of thyroid was confirmed by morphological study. Thus, our data for Ag, Co, Cr, Fe, Hg, Rb, Sb, Sc, Se and Zn (10 trace elements). Our data reveal that there is a significant increase in Co, Rb, Sb and Zn mass fraction in the normal thyroid of female during a lifespan. Therefore, a goitro- and carcinogenic effect of excessive Co, Fe, Rb, Sb and Zn level in the thyroid of old females should be studied.

**Acknowledgement**

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**References**


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**Table 6:** Intercorrelations of the trace element mass fractions in the intact thyroid of female (*r*-coefficient of correlation).

<table>
<thead>
<tr>
<th>Element</th>
<th>Co</th>
<th>Cr</th>
<th>Fe</th>
<th>Hg</th>
<th>Rb</th>
<th>Sb</th>
<th>Sc</th>
<th>Se</th>
<th>Zn</th>
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</thead>
<tbody>
<tr>
<td>Ag</td>
<td>0.182</td>
<td>-0.063</td>
<td>-0.468a</td>
<td>-0.340</td>
<td>-0.231</td>
<td>0.411a</td>
<td>-0.158</td>
<td>0.253</td>
<td>0.012</td>
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<td>1</td>
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<td>-0.043</td>
<td>-0.260</td>
<td>0.572b</td>
<td>0.644c</td>
<td>0.408a</td>
<td>0.608b</td>
<td>0.804c</td>
</tr>
<tr>
<td>Cr</td>
<td>0.499a</td>
<td>1</td>
<td>-0.172</td>
<td>-0.239</td>
<td>0.257</td>
<td>0.319</td>
<td>-0.133</td>
<td>0.213</td>
<td>0.312</td>
</tr>
<tr>
<td>Fe</td>
<td>-0.043</td>
<td>-0.172</td>
<td>1</td>
<td>0.550b</td>
<td>0.108</td>
<td>-0.070</td>
<td>-0.102</td>
<td>-0.240</td>
<td>0.214</td>
</tr>
<tr>
<td>Hg</td>
<td>-0.260</td>
<td>-0.239</td>
<td>0.550b</td>
<td>1</td>
<td>-0.048</td>
<td>-0.351a</td>
<td>-0.150</td>
<td>-0.174</td>
<td></td>
</tr>
<tr>
<td>Rb</td>
<td>0.572b</td>
<td>0.257</td>
<td>0.108</td>
<td>-0.048</td>
<td>1</td>
<td>0.276</td>
<td>0.417a</td>
<td>0.740a</td>
<td>0.581b</td>
</tr>
<tr>
<td>Sb</td>
<td>0.644c</td>
<td>0.319</td>
<td>-0.070</td>
<td>-0.351a</td>
<td>0.276</td>
<td>1</td>
<td>-0.205</td>
<td>0.674c</td>
<td>0.441a</td>
</tr>
<tr>
<td>Sc</td>
<td>0.408a</td>
<td>-0.133</td>
<td>-0.102</td>
<td>-0.083</td>
<td>0.417a</td>
<td>-0.205</td>
<td>1</td>
<td>0.312</td>
<td>0.520a</td>
</tr>
<tr>
<td>Se</td>
<td>0.606b</td>
<td>0.213</td>
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<td>-0.15</td>
<td>0.401a</td>
<td>0.674c</td>
<td>0.312</td>
<td>1</td>
<td>0.470a</td>
</tr>
<tr>
<td>Zn</td>
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<td>0.312</td>
<td>0.214</td>
<td>-0.174</td>
<td>0.581b</td>
<td>0.441a</td>
<td>0.520a</td>
<td>0.470a</td>
<td>1</td>
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</tbody>
</table>


